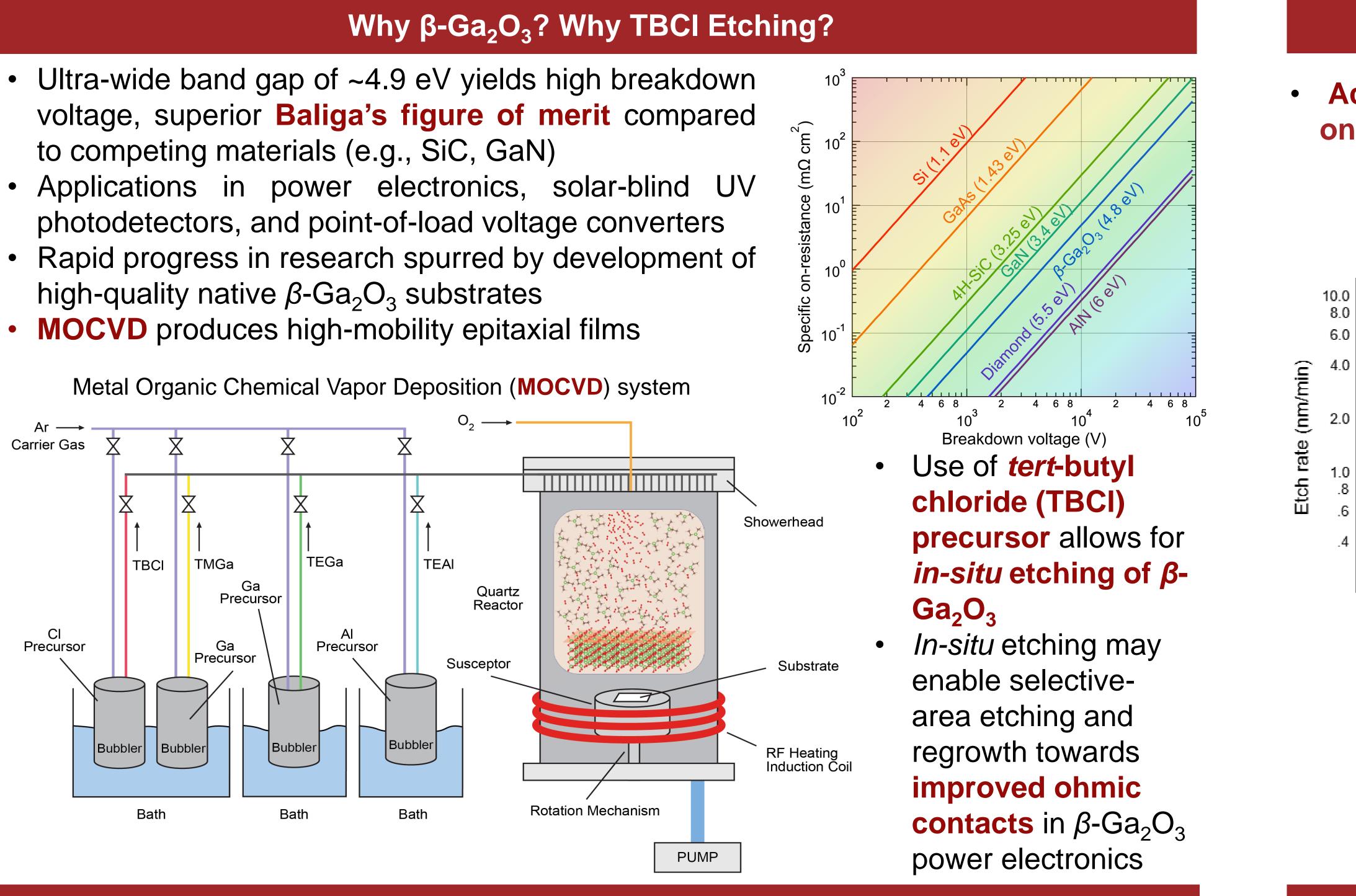
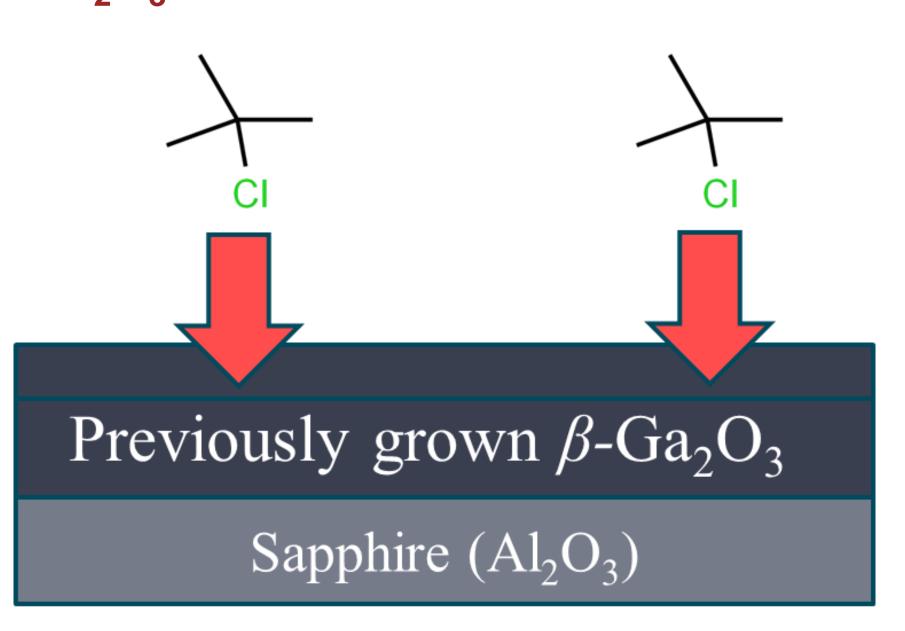


<sup>a</sup>Carleton College Department of Physics and Astronomy. Contact: bowmanh@carleton.edu. 2023 PARADIM REU, Cornell University. Principal Investigator: Professor Hari Nair, Department of Materials Science and Engineering



### How is Etching Controlled?

tert-Butyl Chloride (TBCI) is injected into the MOCVD chamber to etch  $\beta$ - $Ga_2O_3$ 

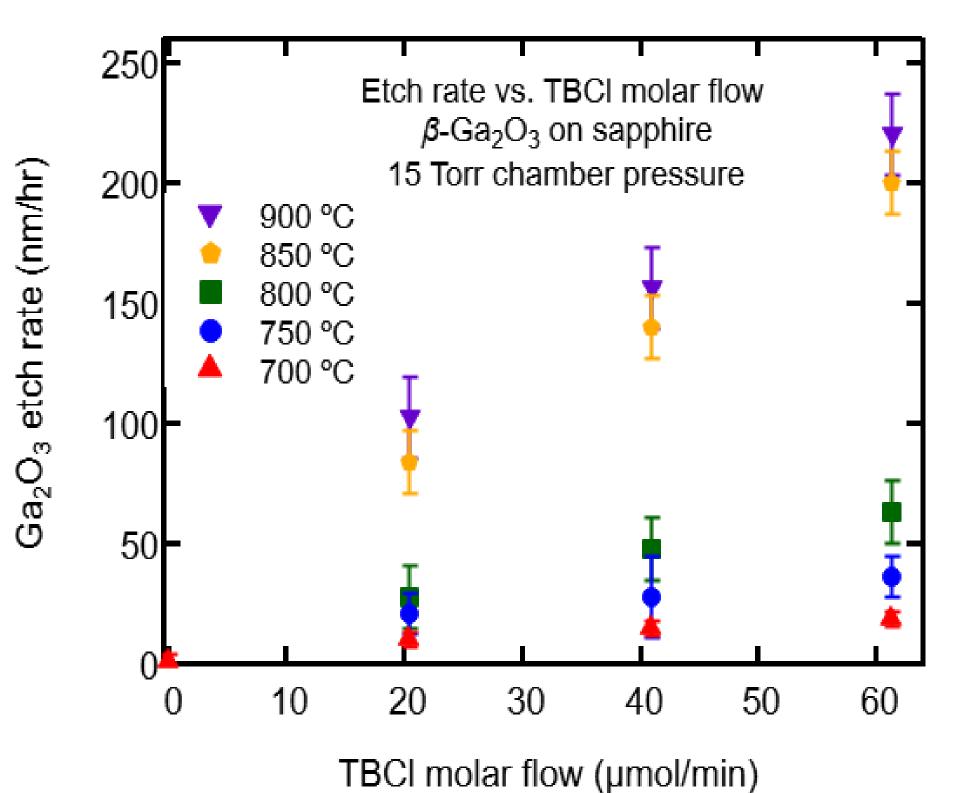


• Heteroepitaxial (grown on c-plane sapphire, shown above) and homoepitaxial  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> (grown on (010) Fe-doped substrates) etched with TBCI

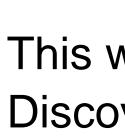
# *In-situ* Etching of β-Ga<sub>2</sub>O<sub>3</sub> Using *tert*-Butyl Chloride in an MOCVD System

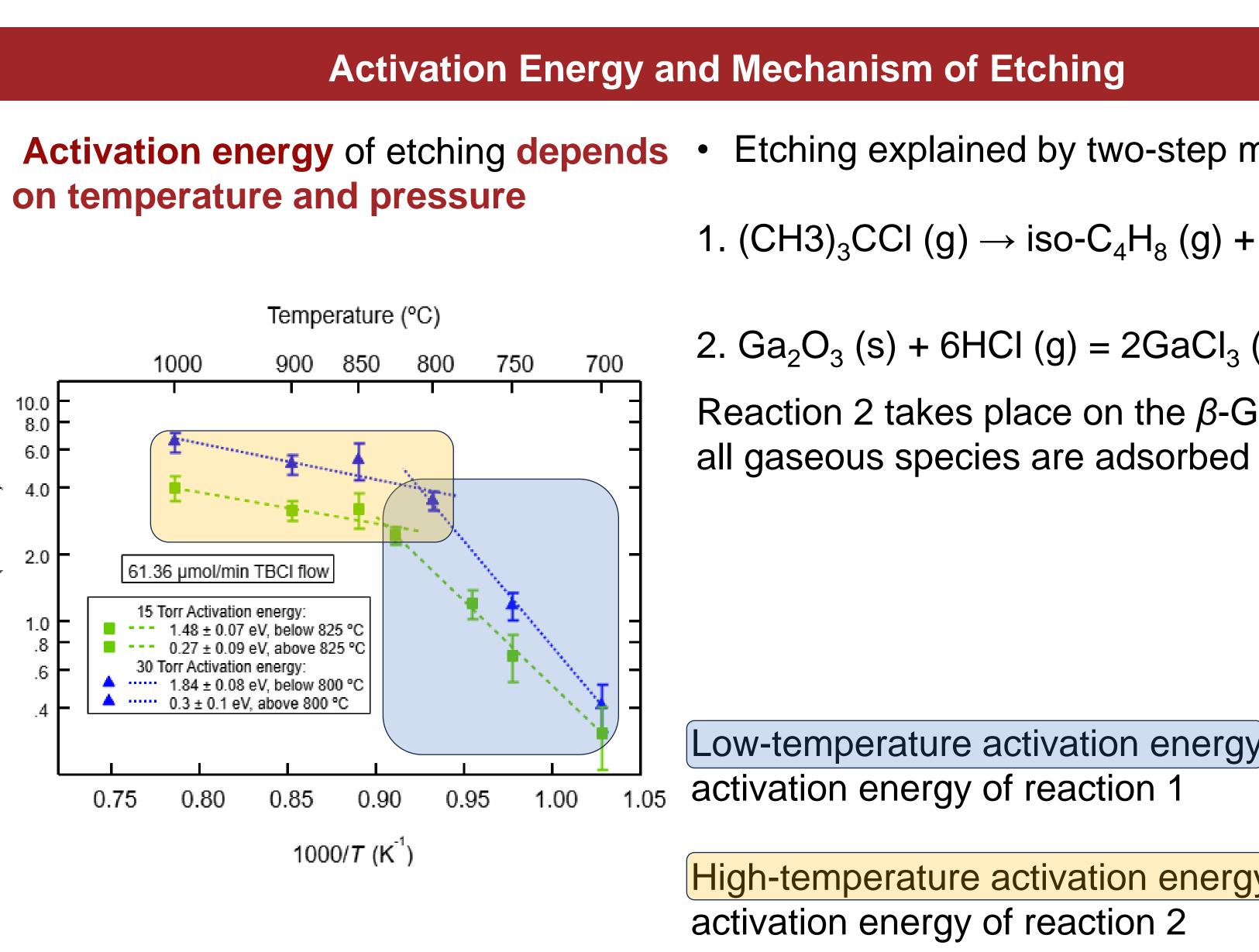
## Henry Bowman<sup>a</sup>

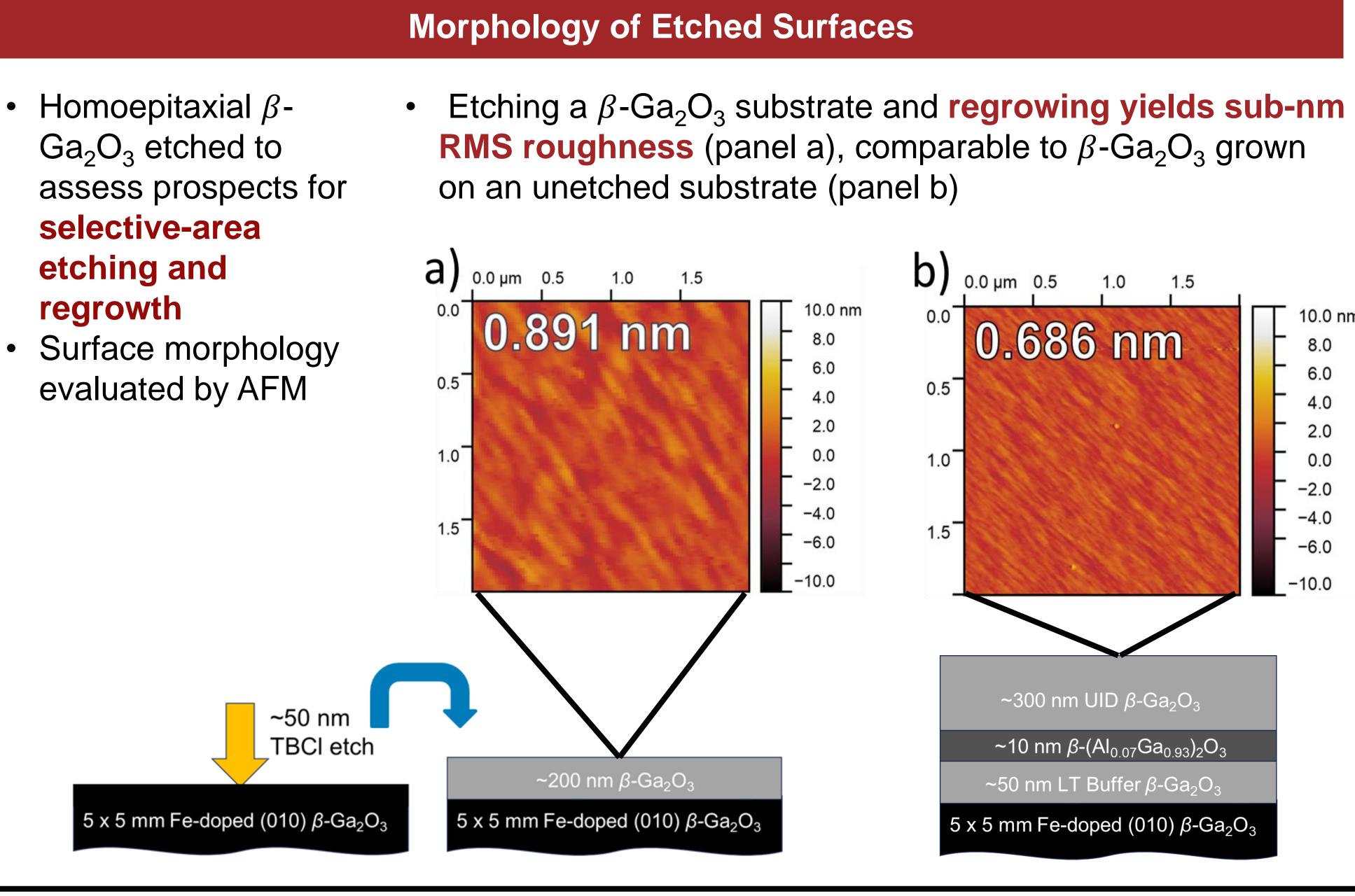
• Etch rate depends on TBCI molar flow and temperature (shown below)



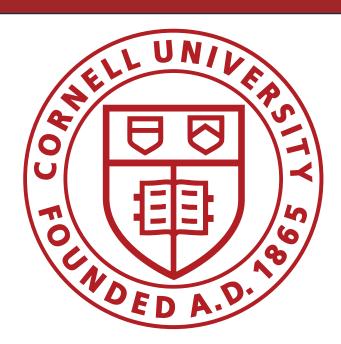
- Etch rate increases with increasing
- TBCI flow and increasing temperature • Etch rate also increases with
- increasing pressure (not shown)







This work was supported by the NSF [Platform for the Accelerated Realization, Analysis and Discovery of Interface Materials (PARADIM)] under Cooperative Agreement No. DMR-1539918.



- Etching explained by two-step mechanism:
- 1.  $(CH3)_3CCI(g) \rightarrow iso-C_4H_8(g) + HCI(g)$
- 2.  $Ga_2O_3(s) + 6HCI(g) = 2GaCI_3(g) + 3H_2O(g)$

Reaction 2 takes place on the  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> surface,

Low-temperature activation energy reflects the

High-temperature activation energy reflects the

