



LECTURE #4— NUTS AND BOLTS OF OXIDE MBE: EPITAXY, SUBSTRATES, AND CRYSTAL GROWTH

Darrell G. Schlom

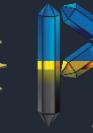
*Department of Materials Science and Engineering
Cornell University*

*Kavli Institute at Cornell for Nanoscale Science
Leibniz-Institut für Kristallzüchtung*

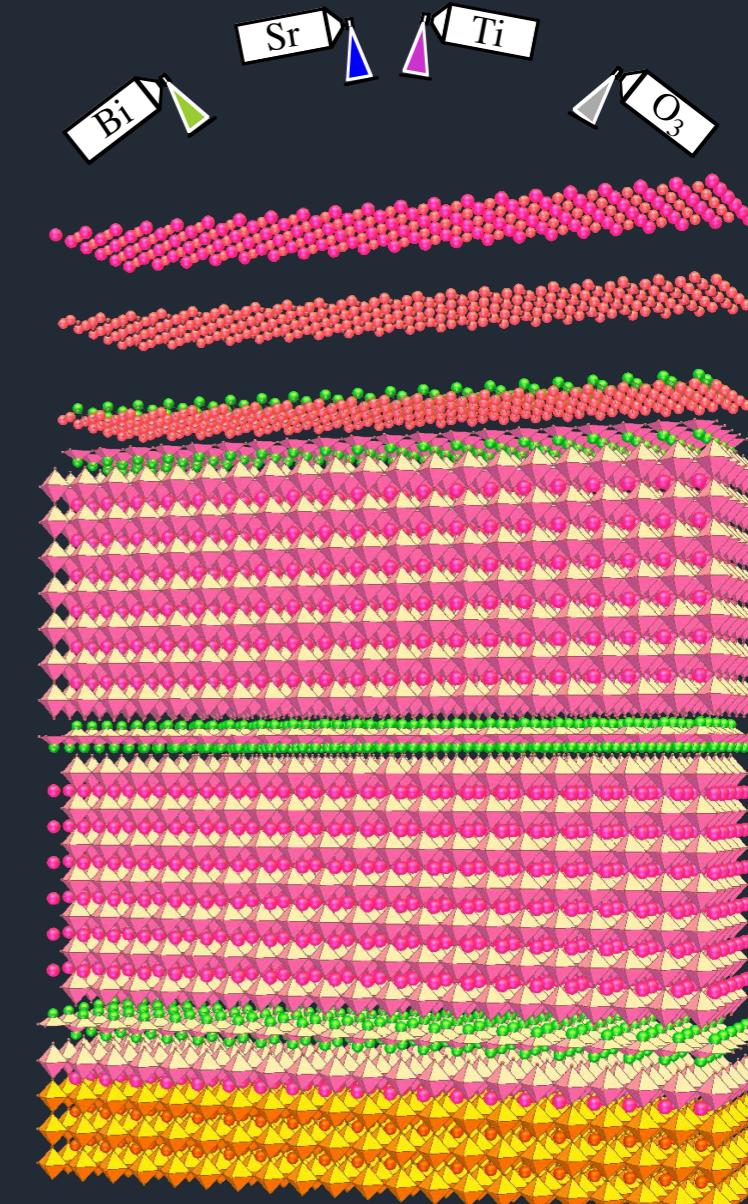
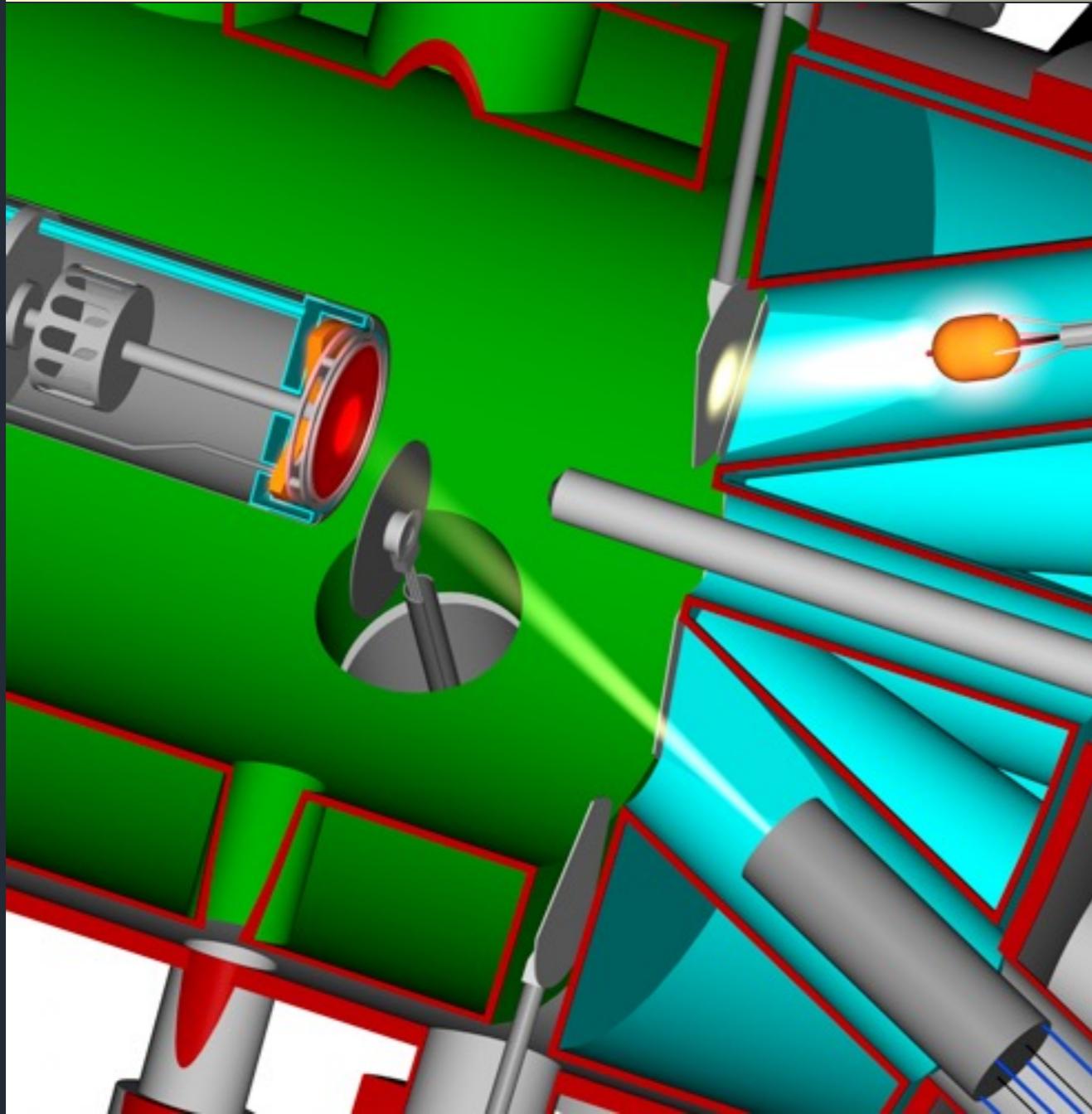
How to grow your favorite oxide by MBE?

- Lecture #2—Growth Conditions, Sources, and Crucibles
- Lecture #3—Composition Control and Calibration
- Lecture #4—Epitaxy, Substrates, and Crystal Growth

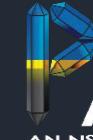
MBE ≈ Atomic Spray Painting



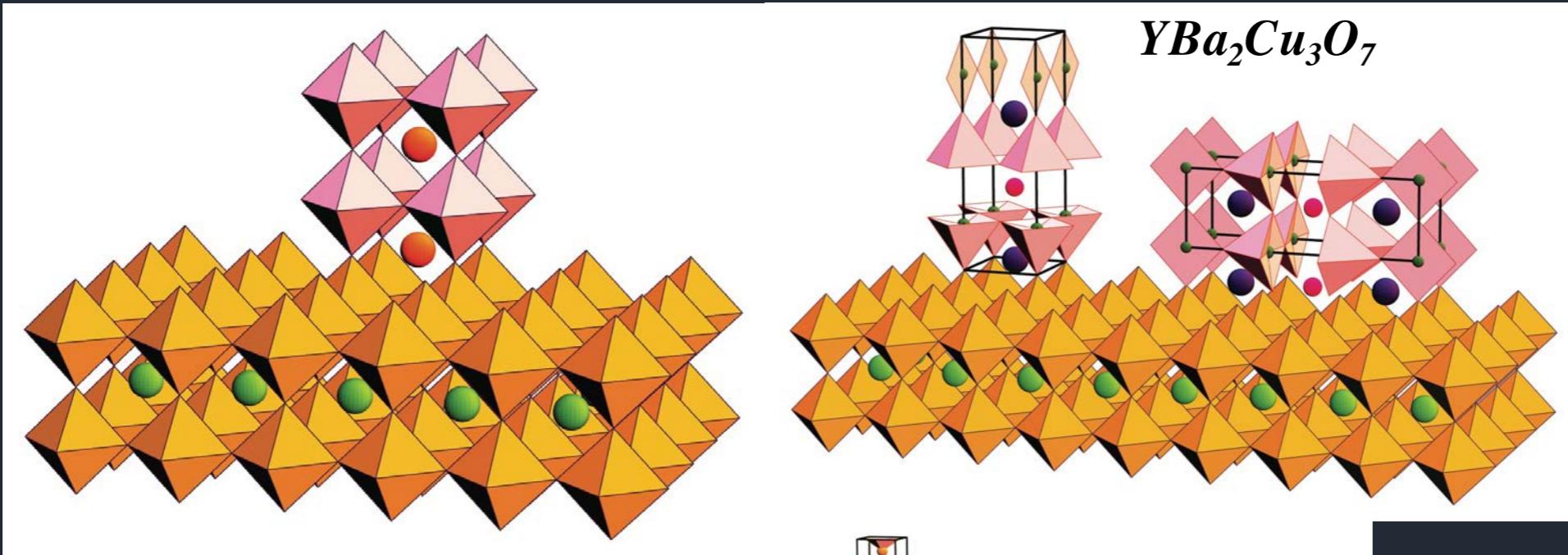
AN NSF MATERIALS INNOVATION PLATFORM



Epitaxial Growth



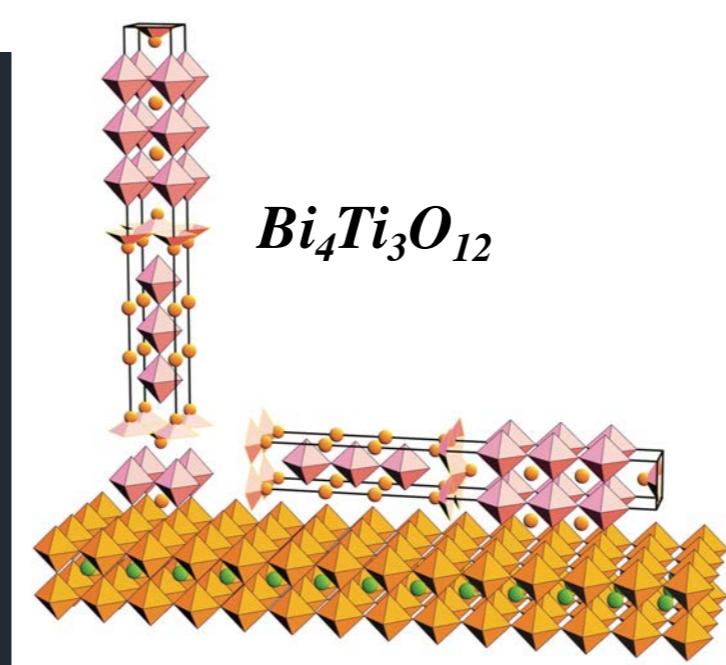
AN NSF MATERIALS INNOVATION PLATFORM



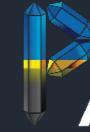
$ABO_3 / A'B'O_3$

“Cube-on-Cube”

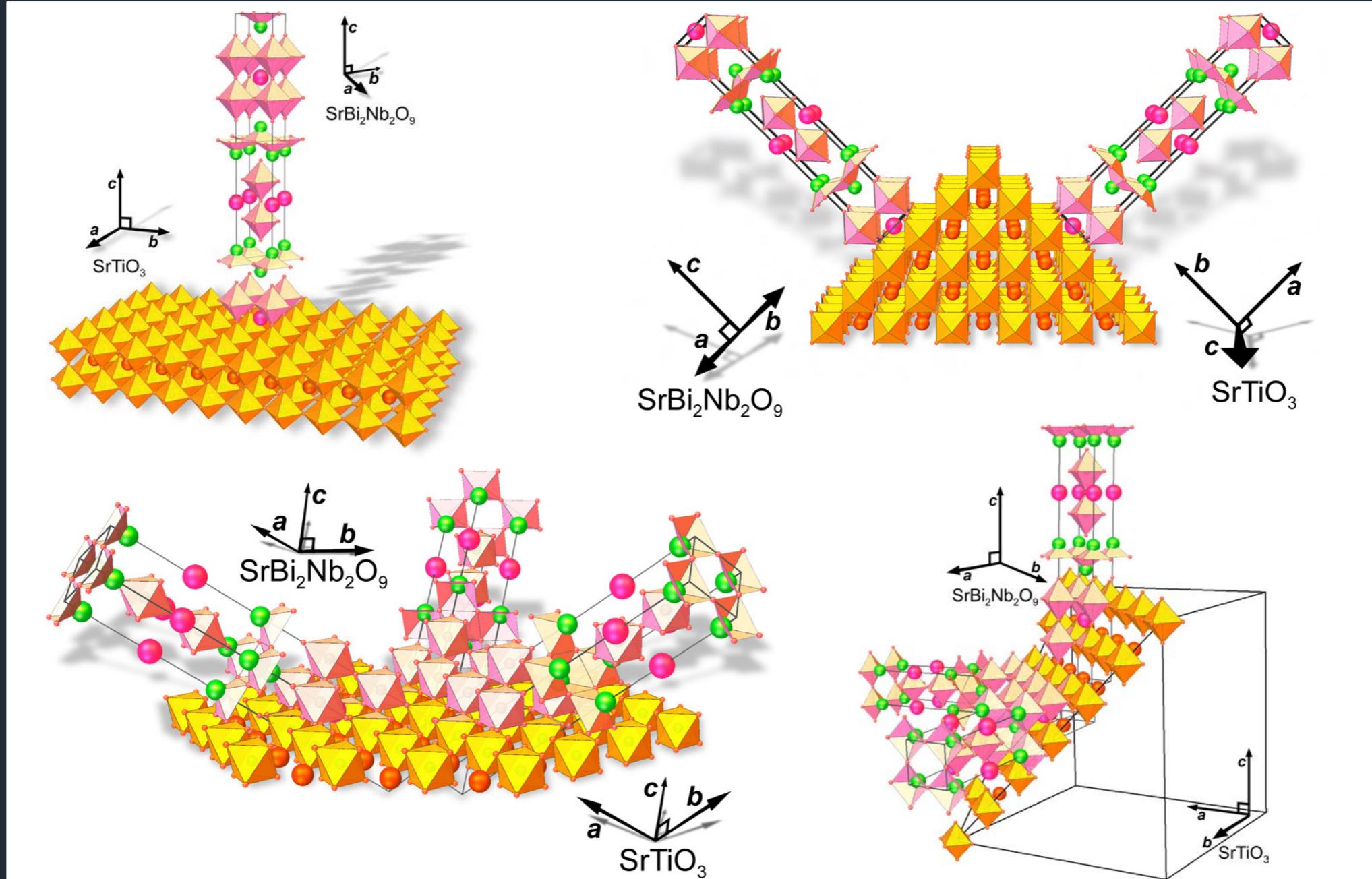
R. Ramesh and D.G. Schlom, *Science* **296** (2002) 1975-1976.



Epitaxial Growth

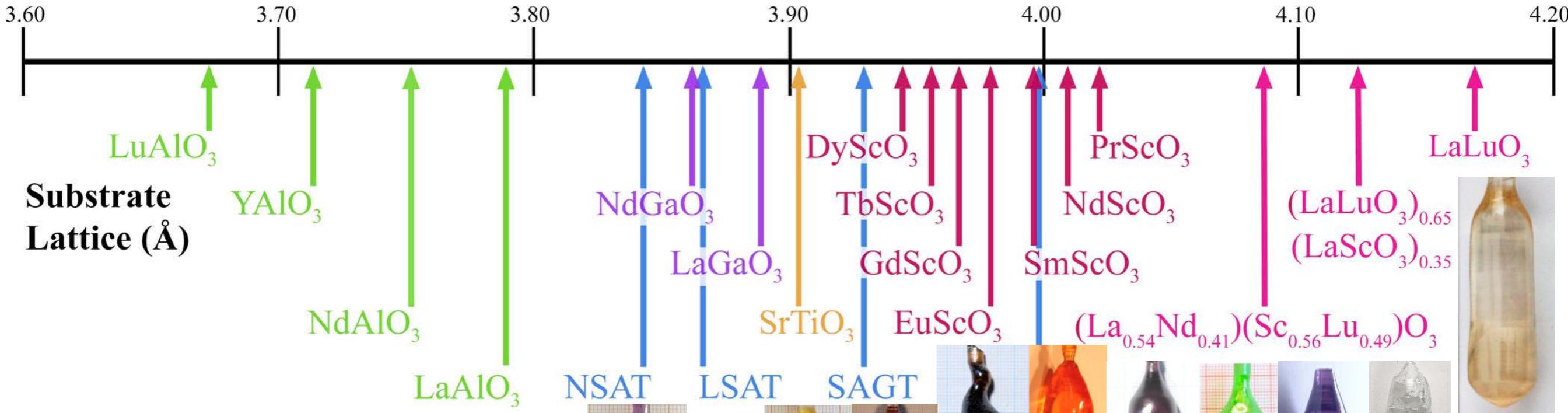


AN NSF MATERIALS INNOVATION PLATFORM



D.G. Schlom, L.Q. Chen,
X.Q. Pan, A. Schmehl, and
M.A. Zurbuchen,
*Journal of the American
Ceramic Society* 91 (2008)
2429-2454.

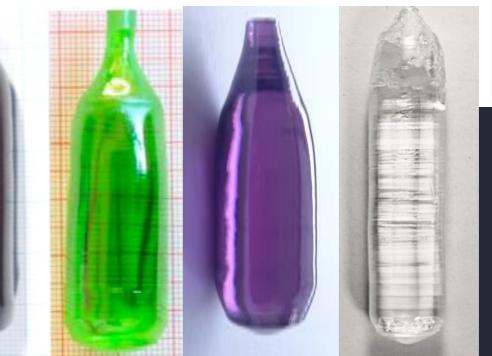
Perovskite Substrates to Impose Strain



Images courtesy of
Christo Guguschev
Institut für Kristallzüchtung
(IKZ)



[110] DyScO₃, $a = 32$ mm



How SrTiO₃ Single Crystals are Grown

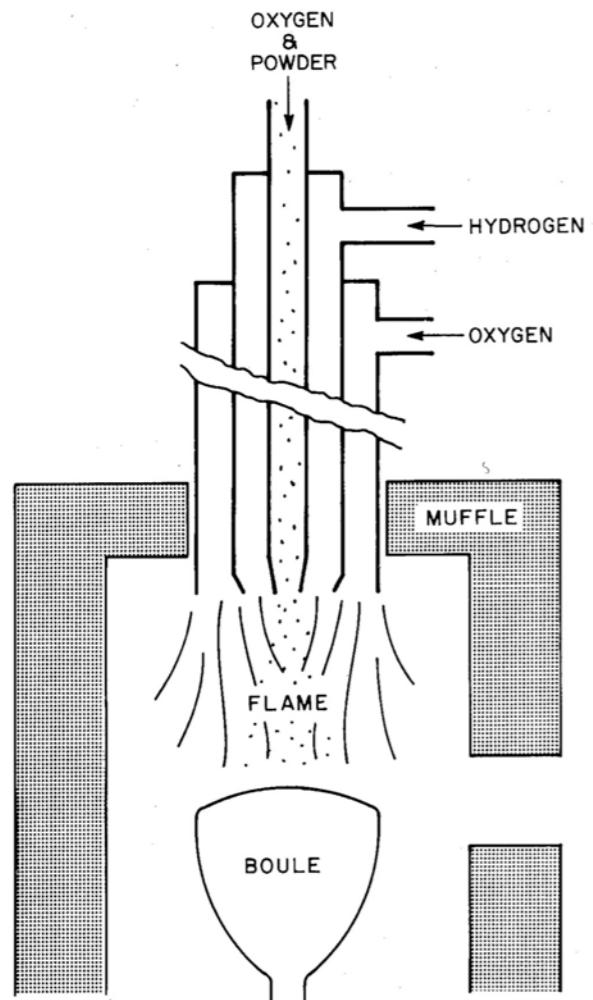
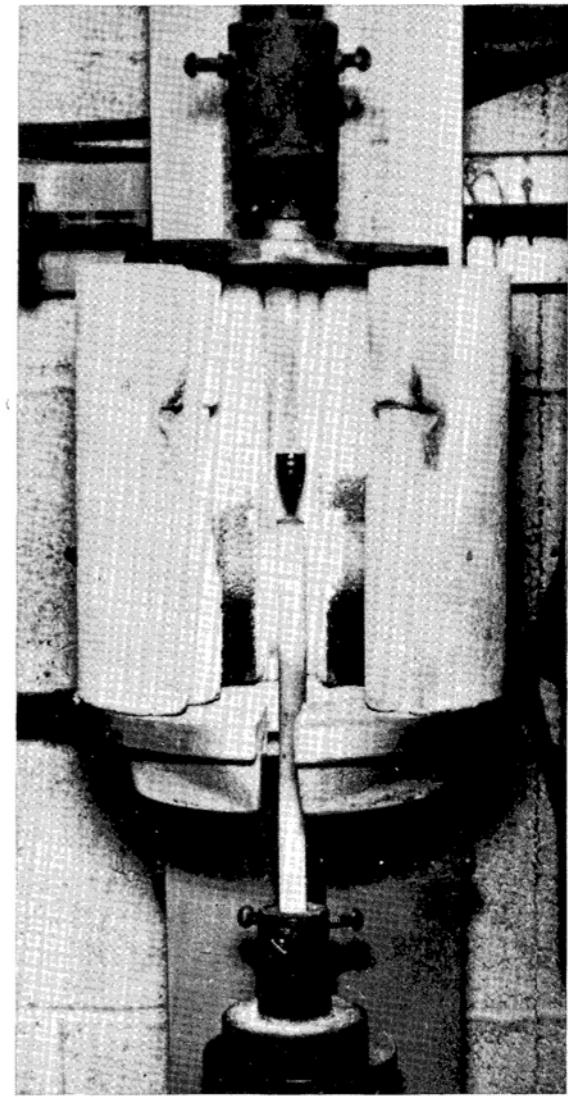


FIG. 18-1. The tricone modification of the Verneuil flame-fusion apparatus used for the growth of synthetic rutile and strontium titanate. The water-cooling arrangement is not shown.

FIG. 18-2. An as-grown black synthetic rutile boule, this will turn a pale yellow after annealing. *Courtesy of N. L. Industries.*



How “Real Crystals” are Grown



AN NSF MATERIALS INNOVATION PLATFORM

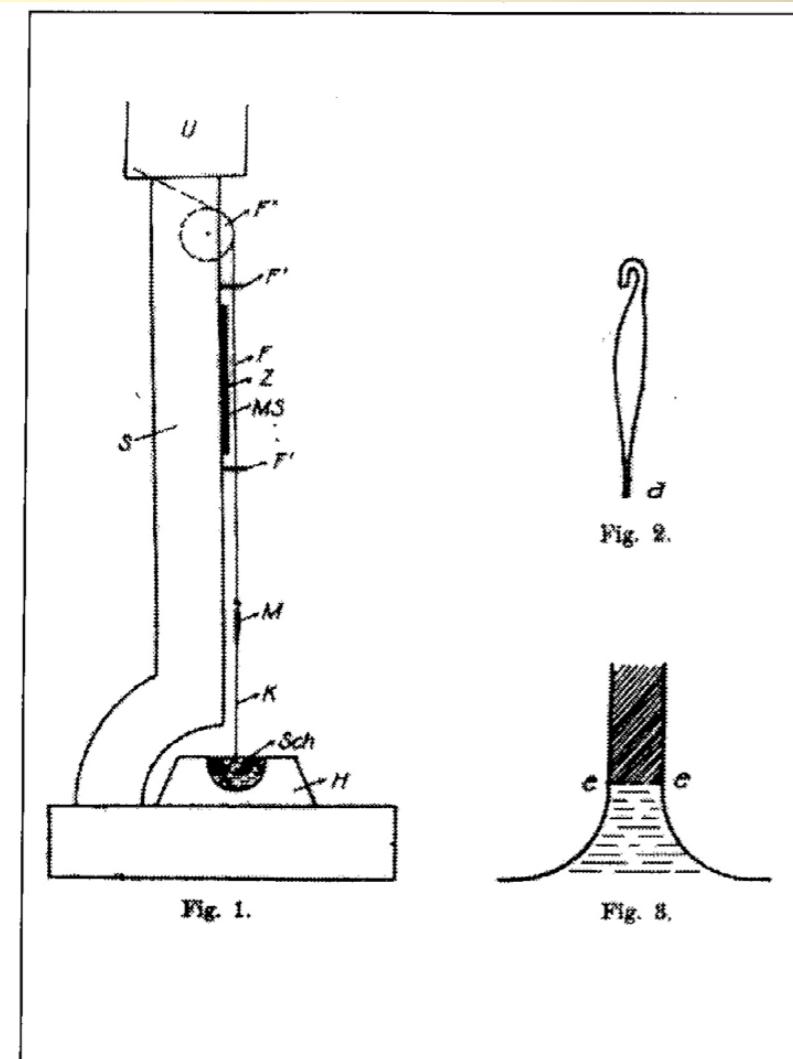


Figure 1. Schematic illustration of Czochralski's method, published in *Zeitschrift für physikalische Chemie* **92** (1918) p. 220.

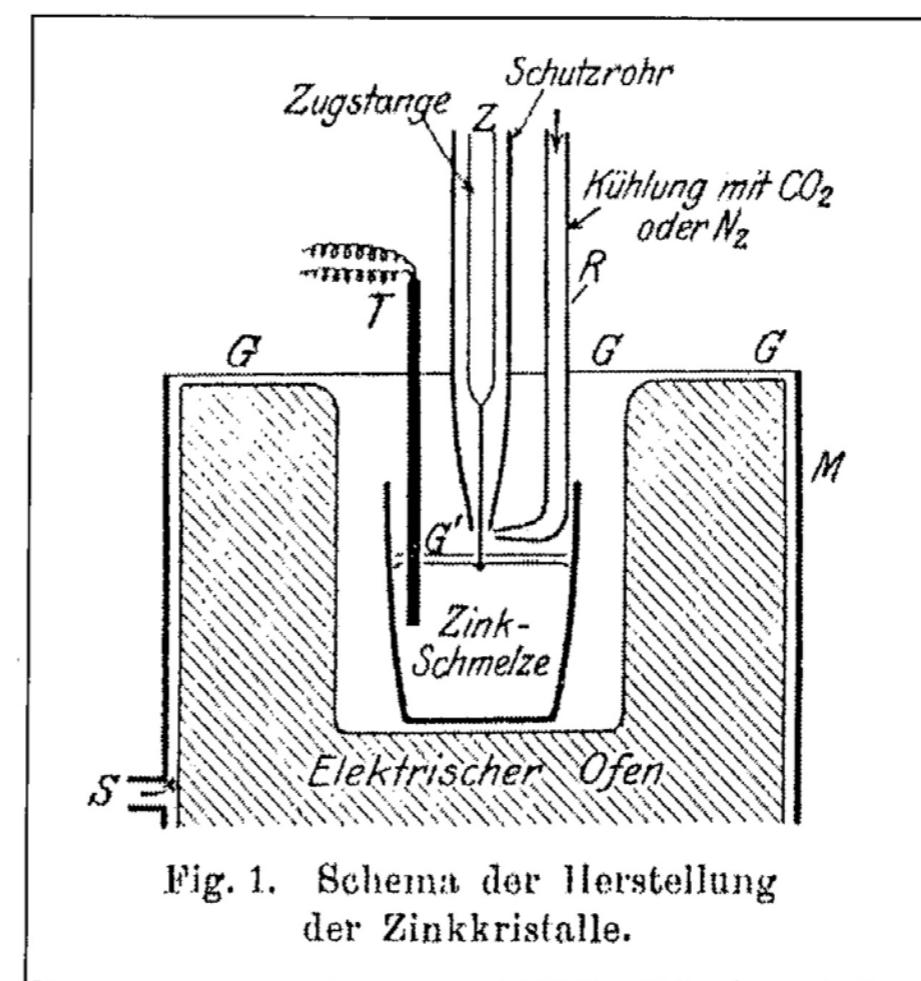
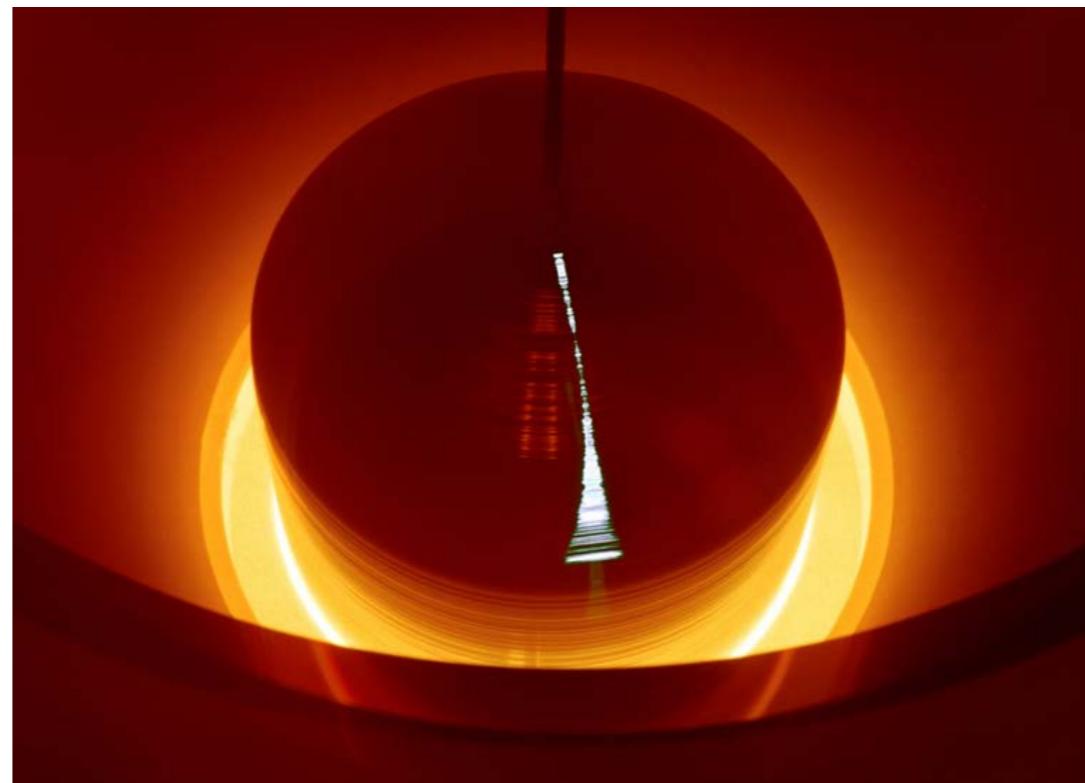
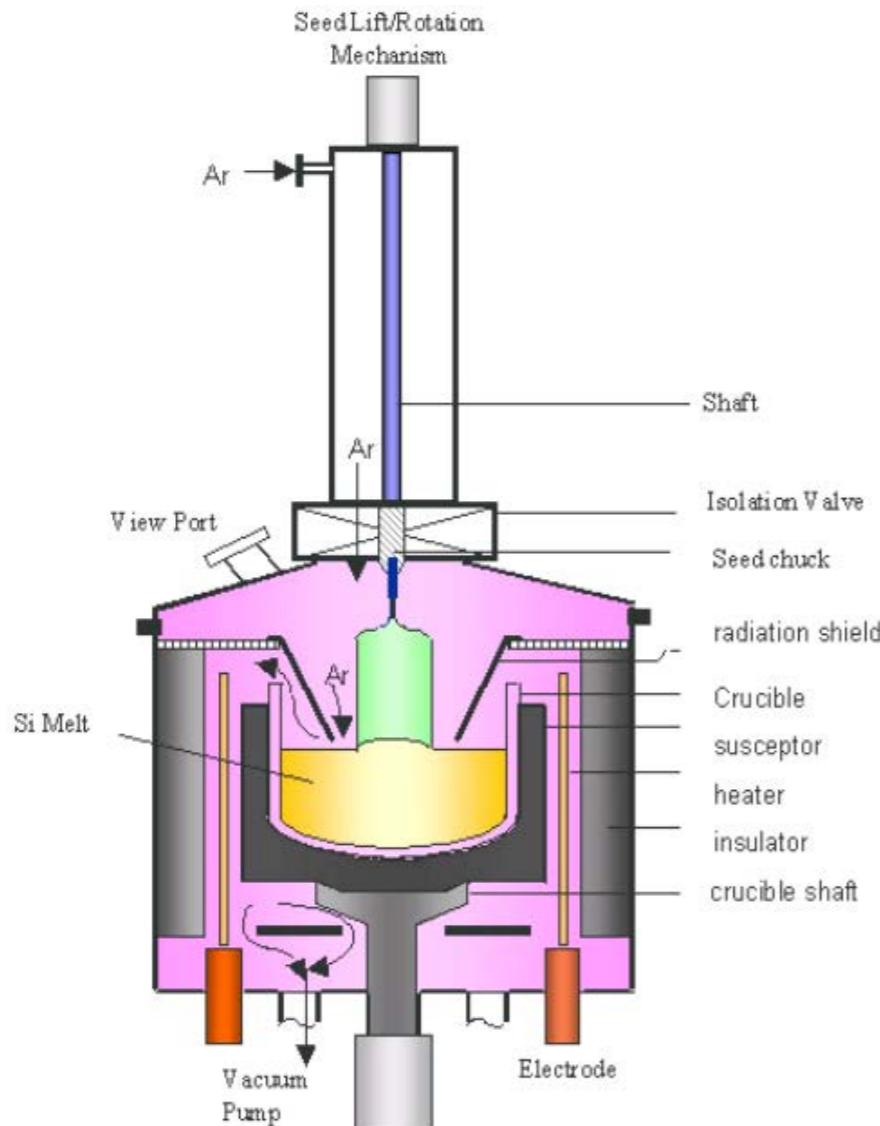


Figure 2. Schematic illustration of Czochralski's method as modified by Mark, Polanyi, and Schmid; published in *Zeitschrift für Physik* **12** (1923) p. 58.

How “Real Crystals” are Grown

CZOCHRALSKI CRYSTAL GROWTH



J.Czochralski, 1916

Surface Termination of Substrates is Important

(100) and (111) SrTiO₃

G. Koster, B.L. Kropman, G.J.H.M. Rijnders, D.H.A. Blank, H. Rogalla,
“Quasi-Ideal Strontium Titanate Crystal Surfaces through Formation of Strontium Hydroxide,”
Applied Physics Letters **73** (1998) 2920-2922.

M. Kawasaki, K. Takahashi, T. Maeda, R. Tsuchiya, M. Shinohara,
O. Ishiyama, T. Yonezawa, M. Yoshimoto, and H. Koinuma,
“Atomic Control of the SrTiO₃ Crystal Surface,”
Science **266** (1994) 1540-1542.

(110) REScO₃

J.E. Kleibeuker, G. Koster, W. Siemons, D. Dubbink, B. Kuiper, J.L. Blok, C-H. Yang,
J. Ravichandran, R. Ramesh, J.E. ten Elshof, D.H.A. Blank, and G. Rijnders,
“Atomically Defined Rare-Earth Scandate Crystal Surfaces,”
Advanced Materials **20** (2010) 3490-3496.

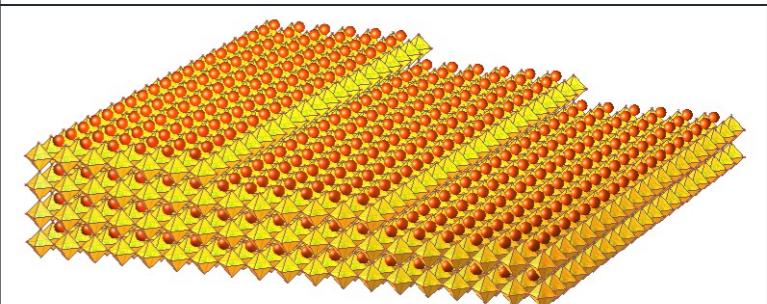
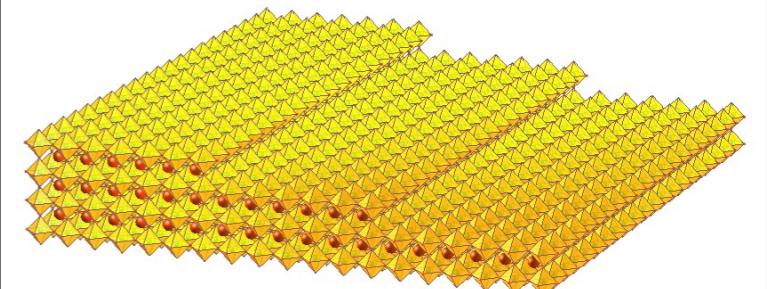
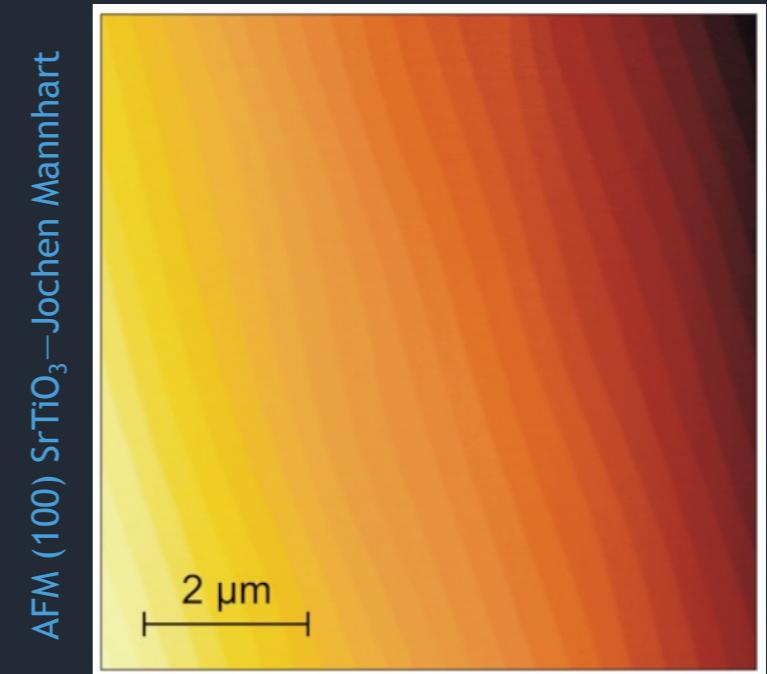
.

(100)_p and (111)_p LaAlO₃

J.L. Blok, X. Wan, G. Koster, D.H.A. Blank, and G. Rijnders,
“Epitaxial Oxide Growth on Polar (111) Surfaces,”
Applied Physics Letters **99** (2011) 151917.

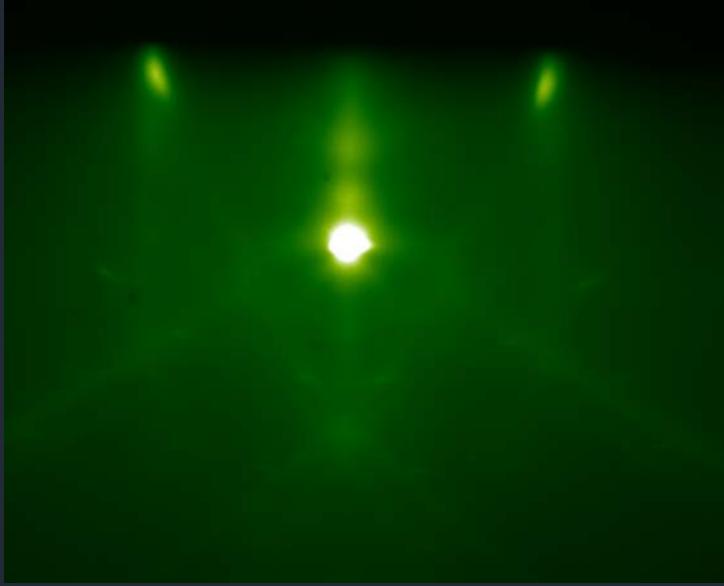
Nice review →

A. Biswas, C.-H. Yang, R. Ramesh, and Y.H. Jeong,
“Atomically Flat Single Terminated Oxide Substrate Surfaces,”
Progress in Surface Science **92** (2017) 117-141.

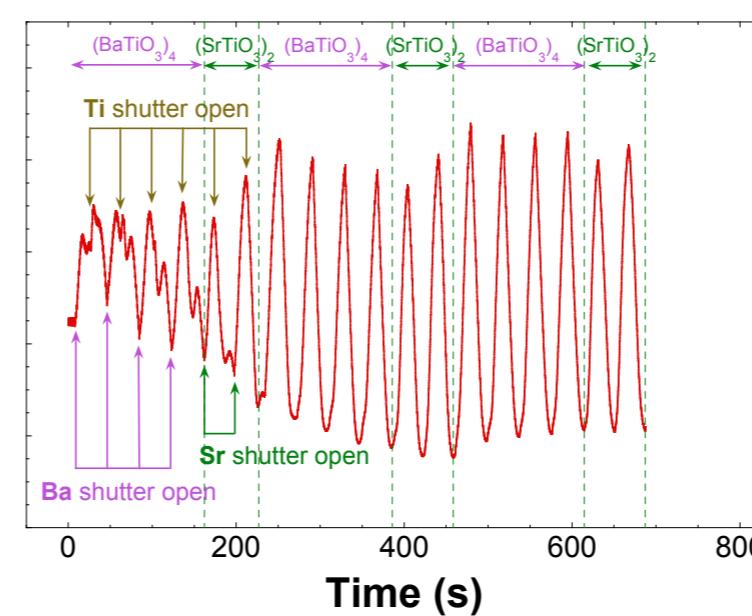


Not Terminated vs. Terminated SrTiO₃

[110] azimuth



RHEED Intensity (arb. units)

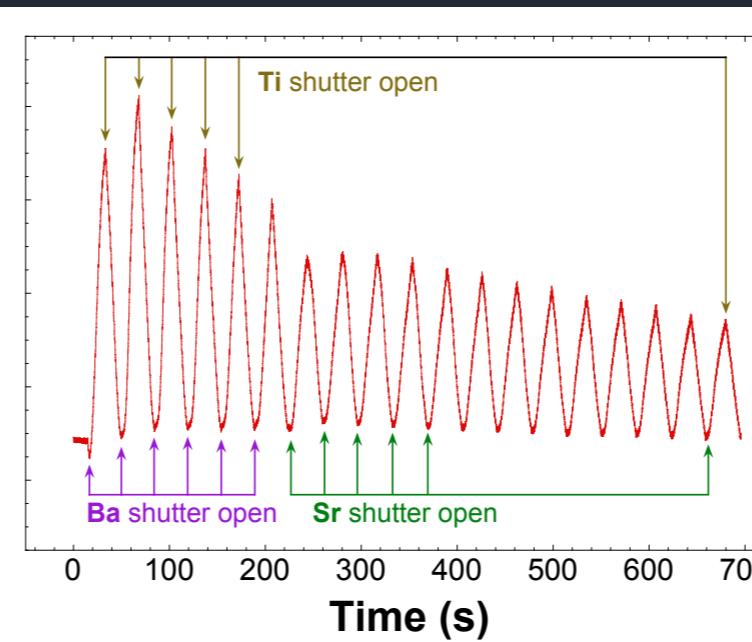


Not
Terminated

[100] azimuth



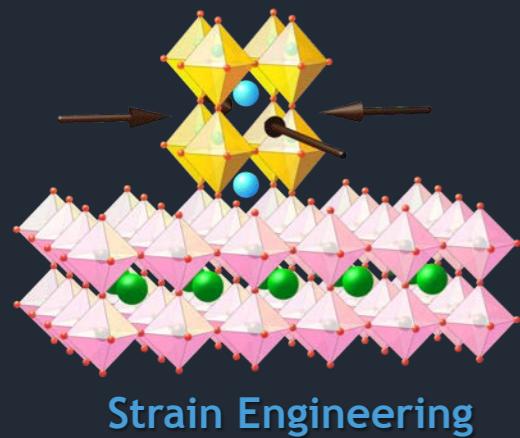
RHEED Intensity (arb. units)



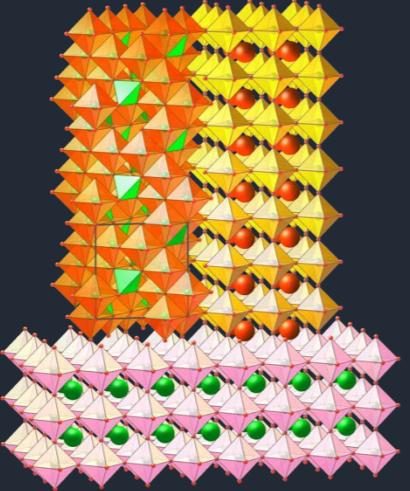
Terminated

A. Soukiassian, W. Tian, V. Vaithyanathan,
J.H. Haeni, L.Q. Chen, X.X. Xi, D.G. Schlom,
D.A. Tenne, H.P. Sun, X.Q. Pan, K.J. Choi, C.B. Eom,
Y.L. Li, Q.X. Jia, C. Constantin, R.M. Feenstra,
M. Bernhagen, P. Reiche, and R. Uecker,
Journal of Materials Research 23 (2008) 1417-1432.

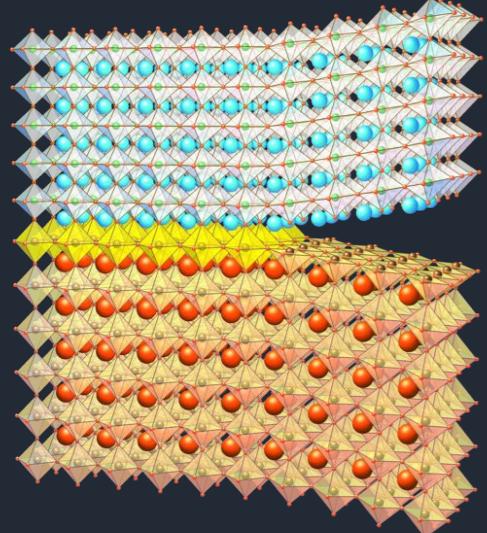
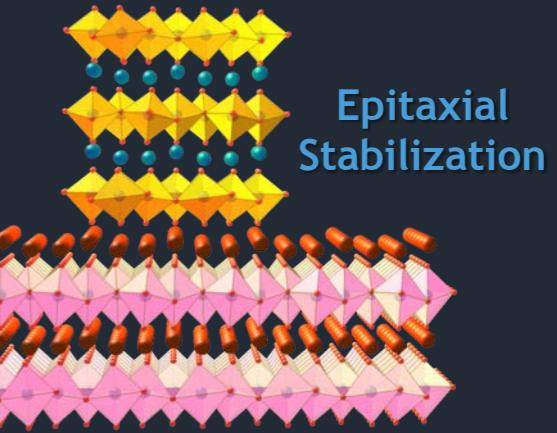
Epitaxial Routes to Engineer Properties



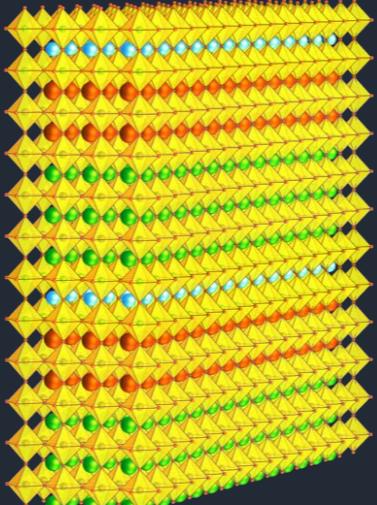
Epitaxial
Nanocomposite



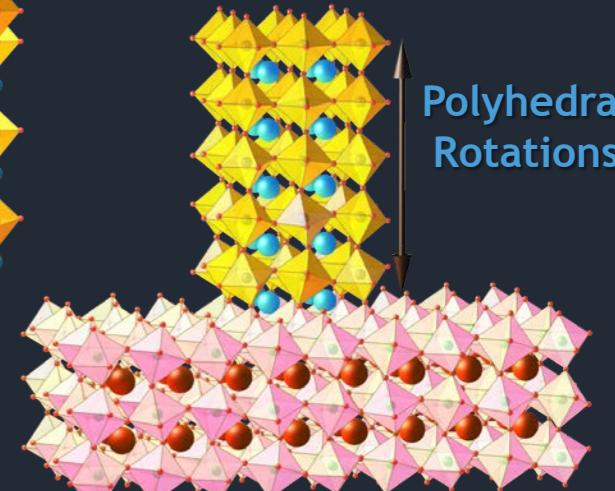
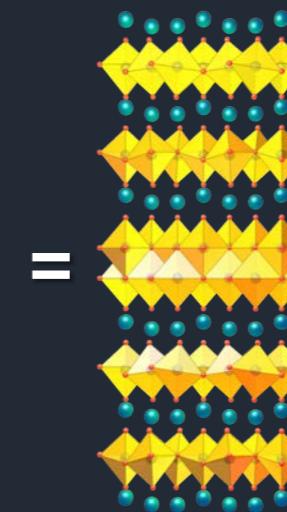
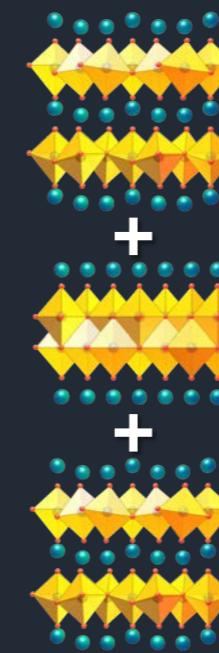
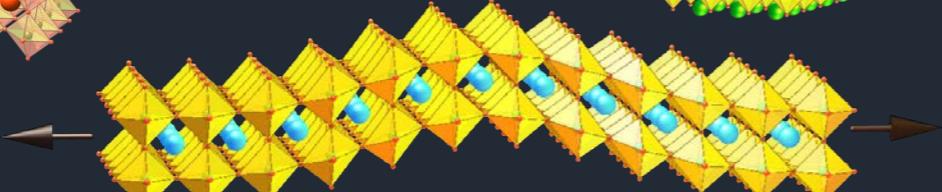
Dimensional
Confinement



Interface
Engineering

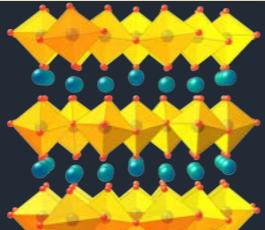


Strained
Membranes



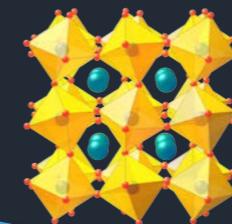
Bulk Thermodynamics

$\Delta G > 0$



versus

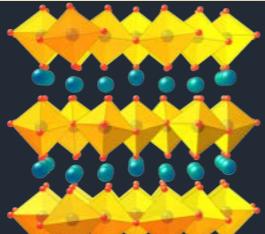
$\Delta G < 0$



Bulk Synthesis

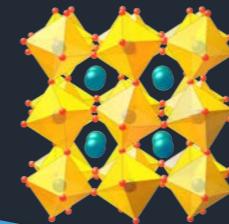
Thin-Film Thermodynamics

$$\Delta G > 0$$



versus

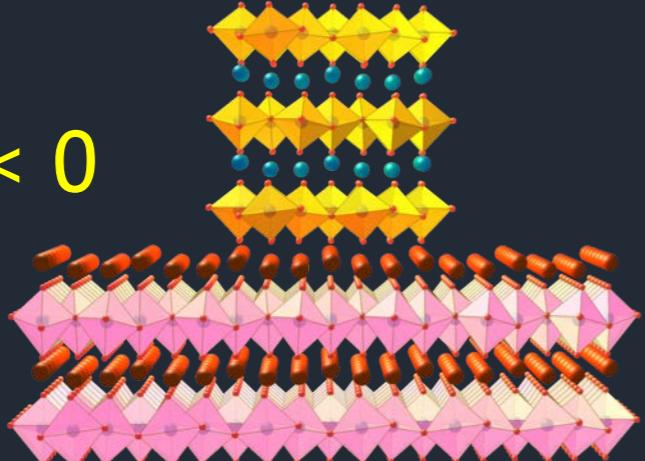
$$\Delta G < 0$$



Bulk Synthesis

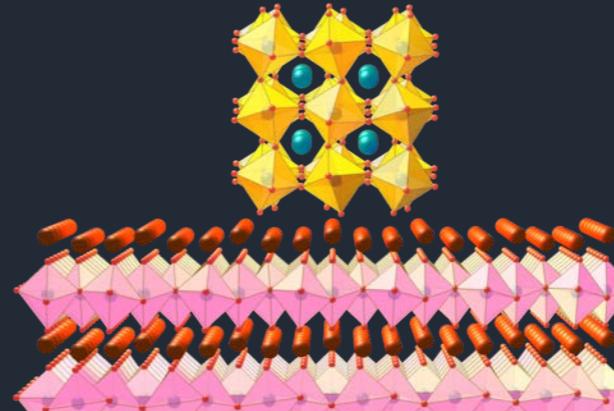
Stable if free energy difference overcome by
 $\Delta(\text{interface energy}) + \Delta(\text{strain energy}) + \Delta(\text{surface energy})$

$$\Delta G_{\text{system}} < 0$$



Epitaxial Stabilization

$$\Delta G_{\text{system}} > 0$$



W.A. Jesser, "A Theory of Pseudomorphism in Thin Films,"
Materials Science and Engineering 4 (1969) 279-286.

Substrates are Key

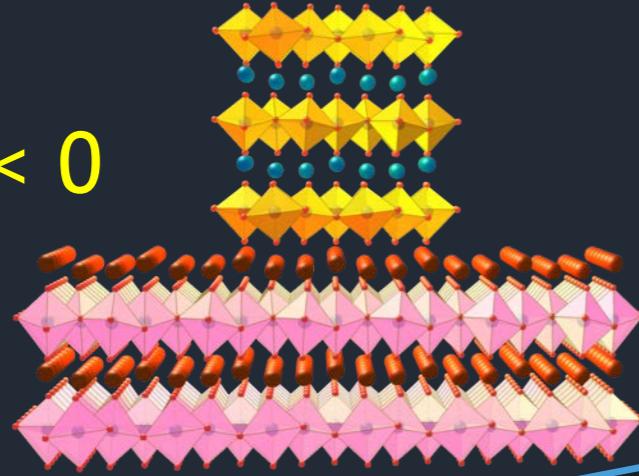
“... the large interface-to-volume ratio found in thin bicrystal films can stabilize the overgrowth in the structure of the substrate rather than its normal bulk structure.”



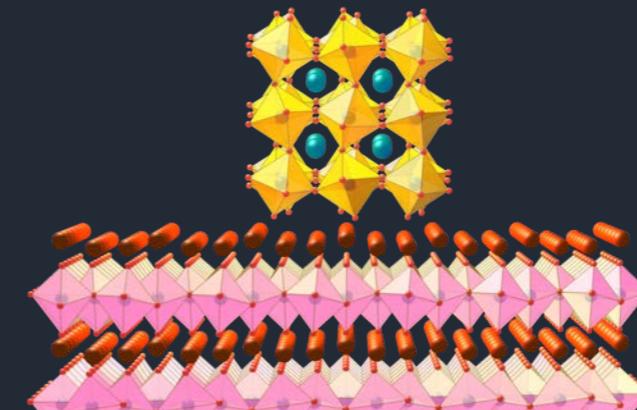
William A. Jesser

Stable if free energy difference overcome by
 $\Delta(\text{interface energy}) + \Delta(\text{strain energy}) + \Delta(\text{surface energy})$

$$\Delta G_{\text{system}} < 0$$



Epitaxial Stabilization



$$\Delta G_{\text{system}} > 0$$

W.A. Jesser, “A Theory of Pseudomorphism in Thin Films,”
Materials Science and Engineering 4 (1969) 279-286.